

A MICROPHOTOGRAPHIC REVIEW OF CORROSION PHENONEMA FROM A SHIPWRECK

The "Rapid", an American China trader, sank approximately 1200 km. north of Perth on January 7th, 1811. She was carrying in her cargo, a large quantity of silver coin. She was later burned to the waterline to conceal her position and her treasure from salvagers.

The Scanning Electron Microscope micrographs exhibit atomic number contrast; That is, lead, a heavy atom, registers as an intensely white area, while calcium or carbon, having a low atomic number, register as dark areas. They were obtained by photographing the SEM TV display using a Nikon SE Automatic camera at F 5.6, 2 second exposure, on Pan F film, 50 ASA. The optical light micrographs were taken using a Zeiss Photo-icroscope microscope II.



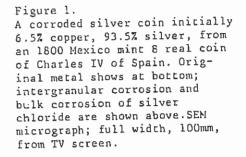




Figure 2. Silver coin showing (from bottom to top): Uncorroded metal band of intergranular corrosion; layers of corrosion products including silver sulphide, cuprous oxide; redeposited copper metal and calcium carbonate (black). The banded nature of corrosion products is due to, in part, major site disturbances when coin was cycled from aerobic to anaerobic to aerobic conditions several times. SEM micrograph; full width, 2.5mm.



Figure 3.
Copper nail concretion showing lower left hand side. Calcium carbonate (CaCO₃) in a Copper Sulphide (CuS) matrix; above the CaCO₃, the white crystal is some lead sulphide; the dark grey is carbonaceous material, probably charcoal and decayed wood. SEM micrograph; full width, 2mm.

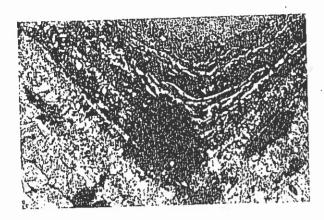


Figure 4.
Copper nail concretion (as
Fig. 3), showing the square
edge of the "nail". The blue
colour represents covellite
(CuS); the brown-grey is a
copper sulphide djurelite,
(approx. Cu 1.96S); the black
material is tenorite (CuO) in
a charcoal matrix. The ship
was burned to the waterline
after being wrecked.
Full width 2.2mm.

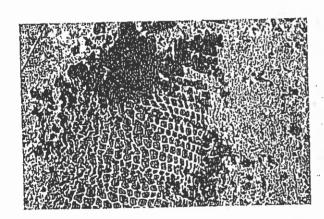


Figure 5.
This is the same source as
Figs. 3 and 4, but showing the
American white oak surrounding
the "nail". The wood cells are
clearly seen, filled with
black-grey chalcocite (Cu₂S)
and surrounded by the blue
covelite (CuS) of the copper
nail concretion.
Full width 1.13mm.

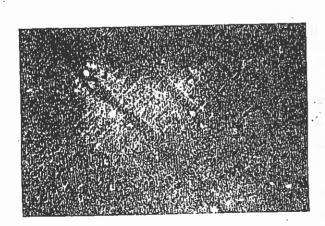
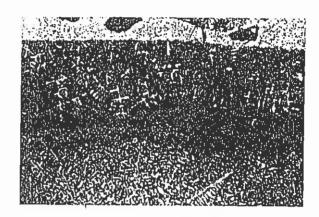


Figure 6.
The subject is the "Rapid's" rudder wood, Larix occidentalis (western larch). The bluegreen at the left hand side is Cu₂(OH)₃Cl as Atacamite, and the pink spots are small crystals of elemental copper corrosion with wood degradation products.
Full width 8.2mm

Figure 7.
Ferric chloride (aqueous), etched, polished specimen from the corroded copper nail.
Upper light-coloured area is copper sulphide (Cu₂S), Chalcocite, with patches of calcium carbonate (CaCO₃) (Black) in it. The area directly below shows residual lead/tin rich phases from the original metal which is lying under a thin band of cuprous oxide. The original metal finished at the Cu₂S residual matrix interface.
Full width, 1.13mm.



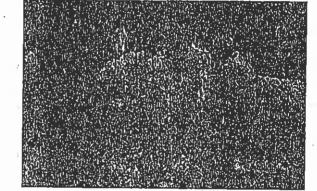


Figure 8.
Polished copper nail showing annealing twins; the nail had been hot worked.
Full width 0.180mm.

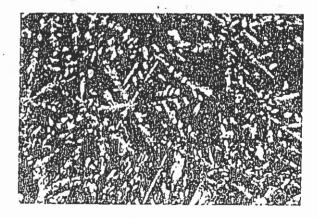


Figure 9.
A polished section of corroded pewter showing dendrites of the tin/antimony intermetallic phase in a tin oxide (SnO₂) cassiterite matrix.
Full width 0.220mm.

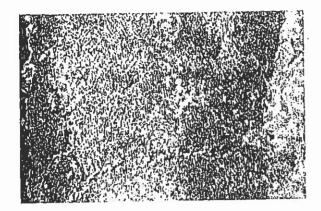


Figure 10.
Copper nail from the bow of the "Rapid". It had been through lead sheet, copper sheet, then into wood. Photo shows residual dendritic metal structure, Cu₂O, cuprous oxide (brown) corrosion products and copper (11) hydroxy chlorides (green) as well as microcracks. In the middle of the picture the light green 'fernlike' pattern is an active 'bronze disease' spot.

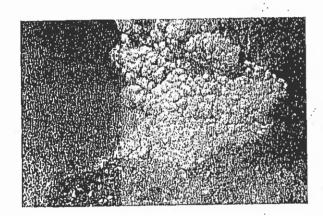


Figure 11.
This photograph can be compared with the above. It is from the leg of the "Bast" on an ancient Egyptian sculpture showing active bronze disease postule.

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17